

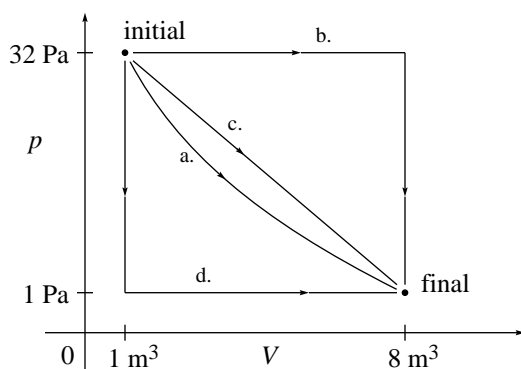
Model Solutions for First Exam

1. The coin toss

A single coin is tossed seven times.

- The probability of obtaining all heads is $1/2^7$.
- The probability of obtaining alternating heads and tails is $2/2^7$. (One alternating pattern starts with heads, the other starts with tails.)
- The probability of obtaining the pattern THHTTHT is $1/2^7$.
- The probability of obtaining a pattern with one head and six tails is $7/2^7$. (There are seven such patterns.)

2. Fluid work



For a quasistatic change, the dissipative work is zero so

$$\text{work} = \text{configuration work} = - \int_{\text{initial}}^{\text{final}} p(V) dV.$$

Now along path (a)

$$p(V) = \frac{K}{V^\gamma} \quad \text{where} \quad K = p_i V_i^\gamma = p_f V_f^\gamma,$$

so the work along path (a) is

$$\text{Work} = - \int_{V_i}^{V_f} \frac{K}{V^\gamma} dV = \frac{K}{(\gamma-1)} \left[\frac{1}{V^{\gamma-1}} \right]_{V_i}^{V_f} = \frac{K}{(\gamma-1)} \left[\frac{1}{V_f^{\gamma-1}} - \frac{1}{V_i^{\gamma-1}} \right] = \frac{p_f V_f - p_i V_i}{\gamma-1}.$$

Path (a) is quasistatic and adiabatic, so $Q = 0$ and $\Delta E = Q + W = W$. Plugging in numbers, $W = -36$ J, $Q = 0$, and $\Delta E = -36$ J. [Note that we never need to calculate any power V^γ .]

Now ΔE is the same for all paths. Thus for paths (b), (c), and (d) we can find W through “negative of area under the path” and Q through $Q = \Delta E - W$. The results are

path	W	Q
a	-36 J	0 J
b	-224 J	188 J
c	-115.5 J	79.5 J
d	-7 J	-29 J

3. Magnetic systems

For these systems (using $x = E/(NH)$)

$$\frac{1}{T} = \frac{\partial S}{\partial E} = \frac{df}{dx} \frac{\partial x}{\partial E} = f'(E/(NH)) \left(\frac{1}{NH} \right)$$

$$\frac{M}{T} = \frac{\partial S}{\partial H} = \frac{df}{dx} \frac{\partial x}{\partial H} = f'(E/(NH)) \left(-\frac{E}{NH^2} \right)$$

Thus

$$M = \frac{M/T}{1/T} = \frac{f'(E/(NH)) \left(-\frac{E}{NH^2} \right)}{f'(E/(NH)) \left(\frac{1}{NH} \right)} = -\frac{E}{H}.$$

This result might be familiar to you from an electricity and magnetism class in the form $E = -MH$.

4. From equation of state to entropy

Combining the two equations in the problem statement gives

$$\frac{p(E, V, N)}{T(E, V, N)} = \frac{\partial S(E, V, N)}{\partial V} = \frac{Nk_B}{V}.$$

If this were a single-variable problem then the right-hand equation would read

$$\frac{dS}{dV} = \frac{Nk_B}{V}$$

and you would immediately integrate this equation as

$$\begin{aligned} dS &= Nk_B \frac{dV}{V} \\ \int dS &= Nk_B \int \frac{dV}{V} \\ S &= Nk_B [\ln V + \text{constant}] = Nk_B \ln(V/V_0) \end{aligned}$$

where in the last step I have written the constant in the form $-\ln V_0$. I prefer this last form because it makes clear that V_0 is a constant with the dimensions of volume.

For the case where S is a function of three variables, the result is exactly the same except the integration over V is carried out at constant E and N , whence the “constant” V_0 , although independent of V , might (and generally will) depend upon E and N . Thus

$$S(E, V, N) = k_B N \ln \frac{V}{V_0(E, N)}.$$

[[You can check for yourself to see that, for the classical monatomic ideal gas, the function $V_0(E, N)$ is given through

$$\frac{1}{V_0(E, N)} = e^{5/2} \left(\frac{4\pi m E}{3h_0^2 N^{5/3}} \right)^{3/2} .]]$$